			20PEB	3122	CHEMISTRY					
Teaching Scheme					Examination Scheme					
L	Т	Р	С	Hours/Week	Theory			Practical		Total Marks
					MS	ES	IA	LW	LE/Viva	1 Otal Walks
3	1	0	4	4	25	50	25	-	-	100

COURSE OBJECTIVES

- > Demonstrate the fundamentals of physical and organic chemistry.
- > Imbibe reaction mechanisms into graduates
- > Improve analytical skills of graduates
- > Improve skills to synthesize chemicals/catalyst.

UNIT I Hours: 10

Physical Chemistry: Reactions and Mechanisms:; Surface Chemistry: Interparticle forces, adsorption isotherms, determination of the surface area fine powders using BET theory, surface films; Colligative Properties and their Experimental Determination: Boiling Point Elevation, Freezing Point depression, Osmotic Pressure

UNIT II Hours: 10

Organic Chemistry:

Organic reactions and their mechanisms: types of organic reactions; general methods of obtaining mechanisms, study of ionic, free radical and other reactions, Alkanes, Alkenes, Alkynes; Markovnikov's rule; Peroxide effect; Bayer's test; Monohydric alcohols; Saytzeff rule; Methods of distinguishing the three classes of alcohols; Aldol condensation; Clemmensen reduction; Wolf-Kishner reduction; Haloform reaction; Cannizzaro reaction; Reformatsky reaction; Wittig reaction; Saturated monocarbozylic acids; Hell-Volhard-Zelinsky reaction; Amino acid; Classification of carbohydrates; Monosaccharides; Mutarotation; Epimerization; Aromatics; Ruff degradation Aromaticity; Huckel rule; Electrophilic substitution reactions; Directive effects of substituents; Aromatic amino compounds; Carbylamines reaction. Organic Geochemistry.

Unit III Hours: 10

Marine Geochemistry: Marine Geochemistry: Ocean as a chemical system: Salinity, Chlorinity, Destiny temperature: ocean circulation and structure of water: Sea water composition; Conservative elements; dissolved gasses, CO₂ distribution etc; Upwelling phenomena; Carbonate dissolution and precipitation; Nutrient elements; Sources and sinks of Dissolved matter and organic Matter; Residence time

Unit IV Hours: 9

Organic Geochemistry and Isotope geochemistry: Organic Geochemistry; A brief Biological background, Organic Compound and their nomenclature, Biologically important organic compounds; Carbohydrate, Lipids, Proteins, Introduction to isotope geology; Discovery or radioactivity and isotopes; Isotope in earth sciences; Nuclide types and their abundances; Decay mechanism of radioactive atoms; basic Principles of radiometric dating; Stable isotope Systematic: elementary knowledge about fractionation of stable isotopes; Stable isotopes and petroleum; mass spectrometry.

COURSE OUTCOMES

On completion of the course, student will be able to

- CO1-: Illustrate the fundamental concepts relevant to organic reaction mechanism.
- CO2- Understand the basic concept of surface chemistry including analysis of BET surface area and colligative properties.
- CO3- Analyse the synthesis and properties of Hydrocarbons (alkane, alkenes and alkynes), monohydric alcohol, monosaccharides, amines, amino acids and aromatic compounds.
- CO4- Demonstrate the mechanism of important name reactions.
- CO5- Explain the basic and fundamental concepts of marine chemistry.
- CO6- Evaluate the principles of organic and isotope geochemistry including important biologically organic compounds.

TEXT / REFERENCE BOOKS

- 1. Finar I.L., "Organic chemistry" Vol-I, 6th Edition, Pearson Education, 2002.
- 2. Sharma B.K., "Industrial Chemistry", 12th Edition, Goel Publishing house, 2001
- 3. Atkins, Peter, 'Physical Chemistry', 8th ed New Delhi : Oxford & IBH Publishing House, 2006.
- 4. Faure G, "Principles of Isotope Geochemistry"
- 5. Killops and Killops, "Introduction to organic Geochemistry"
- 6. White, "organic Geochemistry"
- 7. "Treatise on Geochemistry", 10 volume set, 2006

END SEMESTER EXAMINATION QUESTION PAPER PATTERN

Max. Marks: 100 Exam Duration: 3 Hrs.

PART A: 10 Questions of 2 marks each-No choice

20 Marks
PART B: 2 Questions from each unit with internal choice, each carrying 16 marks

80 Marks